CHALLENGES OF PHYSICO CHEMICAL CHARACTERIZATION OF API

9th Sci & Tech Forum | Markus von Raumer

ACTELION

TABLE OF CONTENTS

- ▶ Drug substance Physical State, Solid Form
- ► Analytical characterization general thoughts and cases



PHYSICO-CHEMICAL CHARACTERIZATION

WHAT IS THE SCALE: MOLECULE VS ENSEMBLE OF MOLECULES

- Examples for molecular properties
 - e.g., pKa, logD

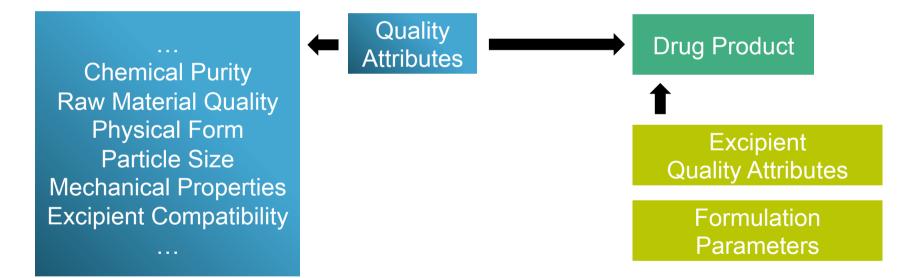
- Examples for ensemble properties
 - e.g., melting point, crystal form

During this presentation physico-chemical characterization will refer to ensemble properties at mostly **room temperature**



FROM DRUG SUBSTANCE TO DRUG PRODUCT

Drug Substance



Not all attributes are necessarily formulation process related



FROM DRUG SUBSTANCE TO DRUG PRODUCT

- ► Influence of the Drug Substance on the Drug Product processing depends on
 - The type of drug product (tablet, capsule, solution, etc.)
 - The concentration of the API (drug load)



DRUG SUBSTANCE - PHYSICAL STATE

BROWN RESINS ...



...WHITE



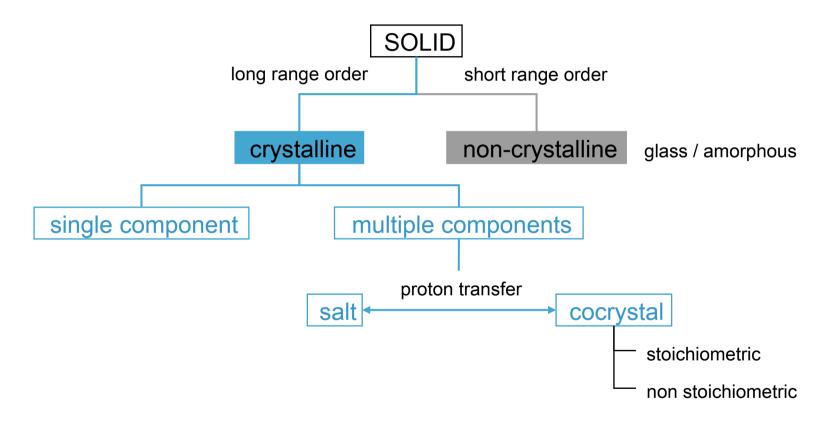




PHYSICAL STATE - HOW TO CLASSIFY SOLIDS

CLASSIFICATION SCHEME OF ORGANIC SOLIDS

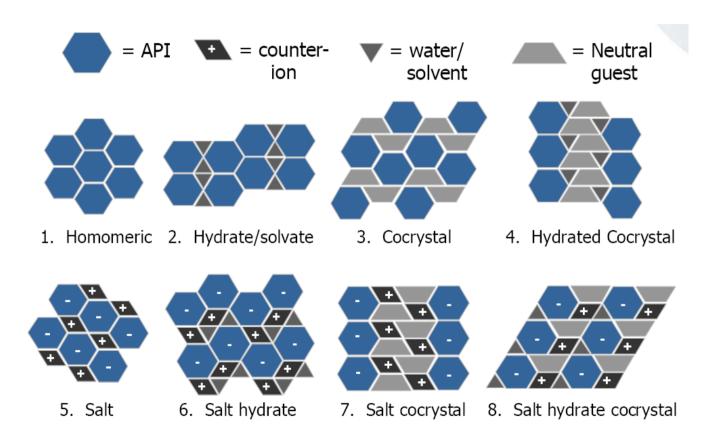
Adapted from: G. P. Stahly, Cryst. Growth Des., 7 (2007) 1007-1026





SOLID FORMS: POSSIBILITIES FOR MULTI-COMPONENT CRYSTALS

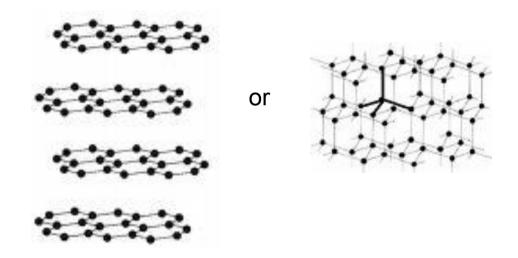
S. Childs, Cocrystallizing pharmaceuticals, presentation at PCL-meeting Bologna 2008





WHAT WOULD YOU LIKE TO HAVE?

Same composition





WHAT WOULD YOU LIKE TO HAVE?

Same composition



graphite



diamond



or

POLYMORPHISM

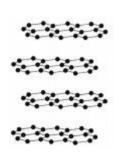
Called allotropism when exhibited by elements.

► Example: carbon

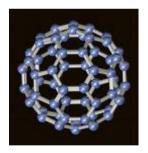
Same composition, different spatial arrangement, ...



diamond



graphite



buckminsterfullerene

... different properties and value



POLYMORPHISM

EXAMPLE FOR SMALL ORGANIC MOLECULE

Yu et al.

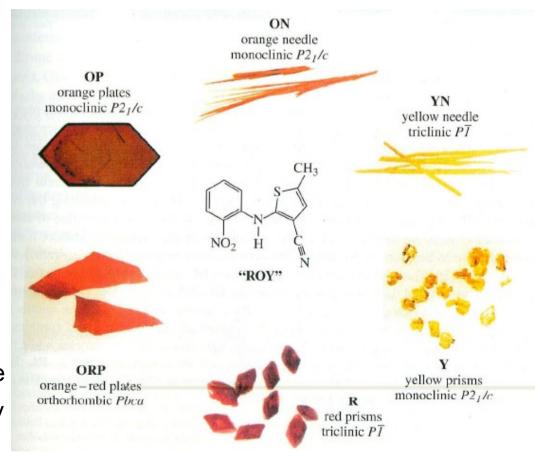
J. Am. Chem. Soc. 122, 585-591 (2000)

J. Phys. Chem. A106, 544-550 (2002)

ROY (Red Orange Yellow)

- ▶ Visual differences in
 - color
 - shape
- ▶ Phys.chem properties also change
 - melting, stability, hygroscopicity

– ...





WHY BOTHER?

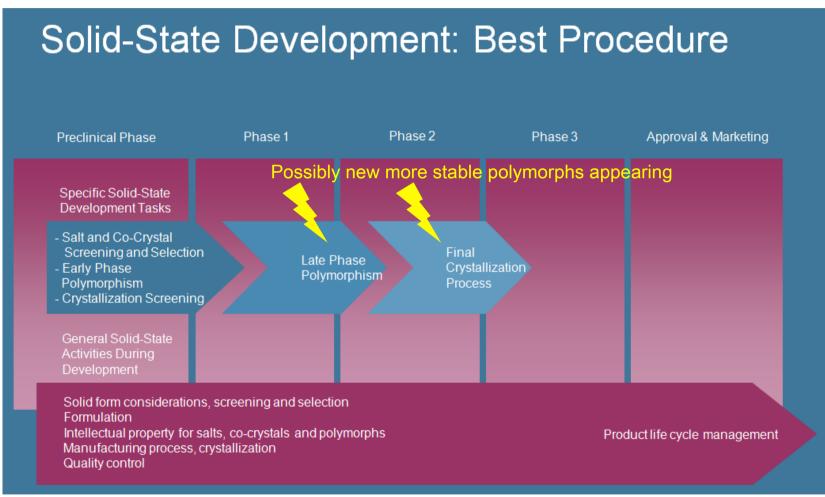
PHYSICO-CHEMICAL PROPERTIES!

- ► The properties of different solids are different, e.g.:
 - Melting point
 - Chemical stability
 - Moisture sorption
 - Mechanical properties (e.g., plasticity, ...)
 - Habit, flow
 - ...
- ▶ In turn, by screening you can tune and possibly optimize selected properties



DRUG SUBSTANCE

TYPICAL PROCEDURE IN VIEW OF PHARM, DEV. - TIME

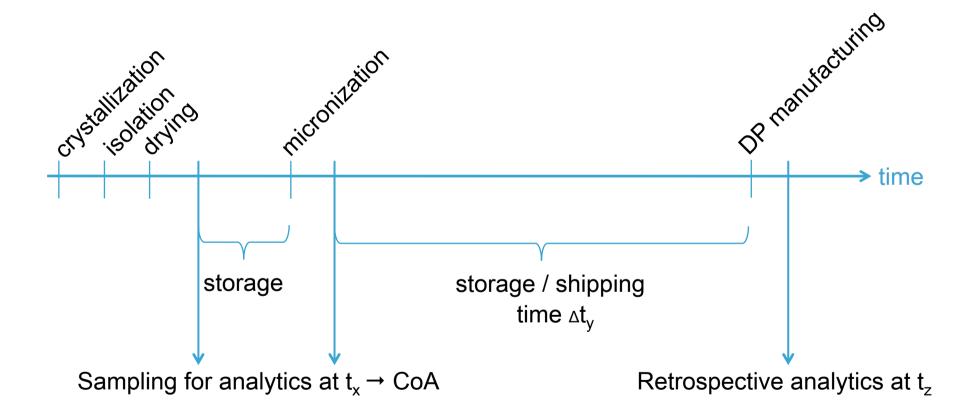


Solvias AG



DRUG SUBSTANCE

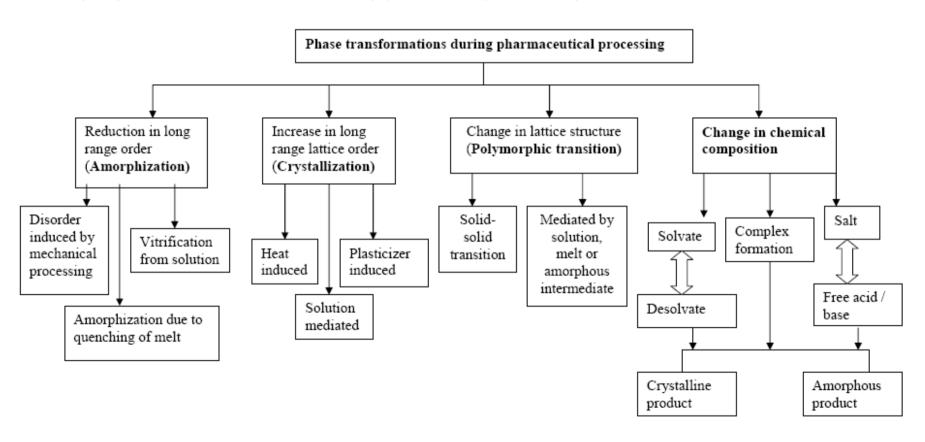
TYPICAL PROCEDURE: TIME AS A VARIABLE





WATCH AND UNDERSTAND YOUR PROCESS!

R. Govindarajan and R. Suryanarayanan 'Processing-induced phase transformations' in: Polymorphism in the Pharmaceutical Industry (R. Hilfiker ed.), 2006, Wiley-VCH





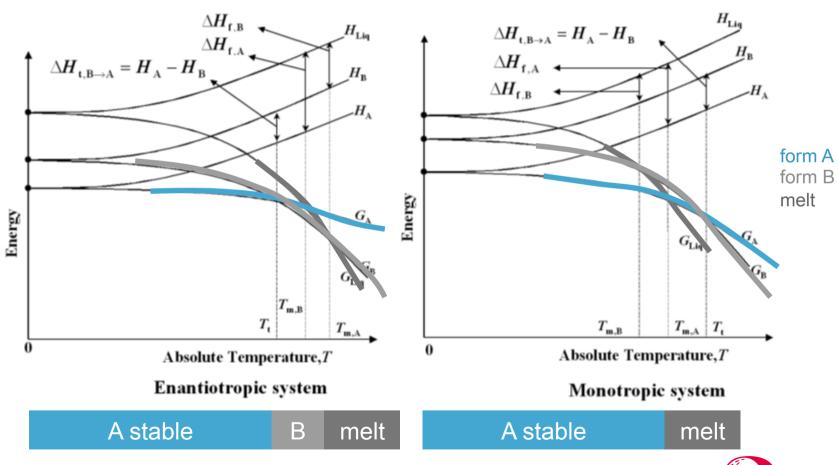
CRITICAL?

- Is the attribute that you measure at defined conditions relevant of the process conditions?
- Drug substance solid characteristics must be understood!
 - E.g., polymorphism monotropism / enantiotropism
 - Multicomponent systems
 - Salts: disproportion as function of microenvironmental pH Merritt et al., 'Implementing QbD in Pharmaceutical Salt Selection: A modeling approach to understanding disproportionation', Pharm Res 30, 2013, 203-217
 - **Hydrates**
 - Behaviour as function of r.h at constant temperature
 - Behaviour as function of temperature
 - Behaviour in function of excipients



POLYMORPHISM: ENANTIOTROPISM & MONOTROPISM

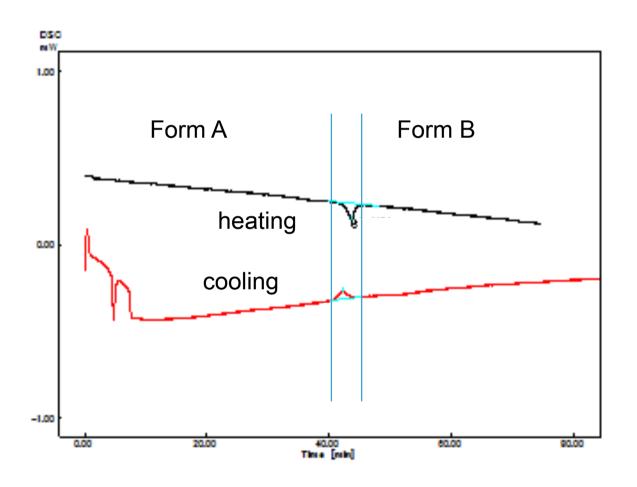
KNOW YOUR SYSTEM





POLYMORPHISM: ENANTIOTROPISM

EXAMPLE: RAPID REVERSIBLE SOLID-SOLID TRANSFORMATION



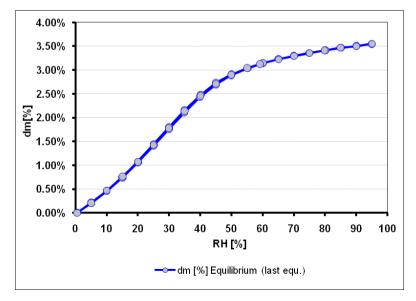


DRUG SUBSTANCE CHARACTERISTICS: HYDRATE

Widely described in literature

S.M. Reutzel-Edens & A.W. Newman, 'Physical Characterization of Hygroscopicity in Pharmaceutical Solids' in: Polymorphism in the Pharmaceutical Industry (R. Hilfiker ed.), 2006, Wiley-VCH

► Example of DS: Probing by gravimetric vapor soprtion (GVS)

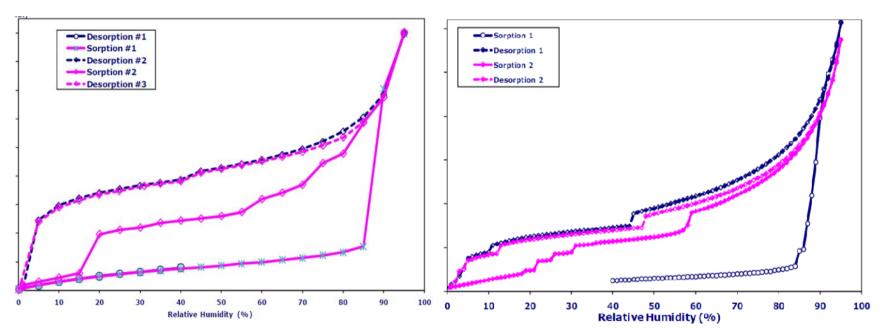




DRUG SUBSTANCE CHARACTERISTICS: HYDRATE

► Constant temperature

Time and step size matter!



Standard procedure: 5% step size

Refined procedure: 1% step size



DRUG SUBSTANCE CHARACTERISTICS: HYDRATE

▶ Peritecticum

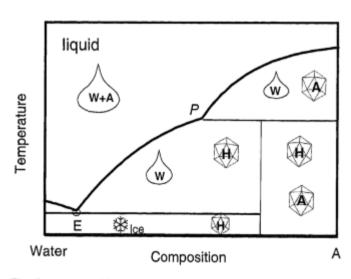
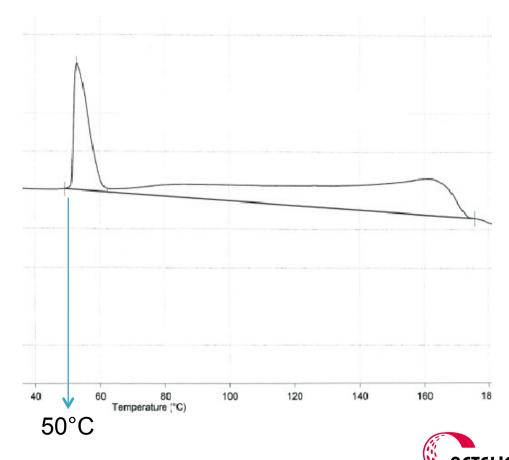


Fig. 5. A composition versus temperature (x-T) phase diagram of ater and compound (A) showing an incongruently melting 'rate (H) form. The hydrate form is stable between the eutectic nd the peritectic (P) points below the liquidus curve.

K. Morris, Adv. Drug. Del. Rev., 48, 91-114 (2001)



TYPICAL PROCEDURE

- Certificate of analysis (CoA)
 - primary focus on purity and aspect criteria
 - Some physico-chemical information (for information and case dependent),
 e.g.:
 - Melting
 - XRPD
 - Particle size distribution
- ► Methods evolve with time of development
 - Generic vs specific methods
 - Validation status



DILEMMA

- In early phase critical quality attributes are not yet known
- ► It makes sense to broadly characterize early drug substance batches beyond methods that are then put on the CoA
 - Implement a batch tracking strategy until the DS → DP link is understood (or thought to be understood)
 - How to sample? When to sample?
 - Aging?
 - What techniques, what method, what granularity?
- Collect data

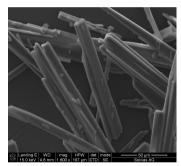


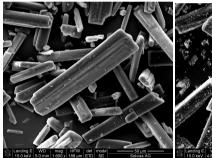
EXAMPLE (1-1)

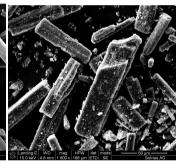
- Upon scale-up of DS poor flow of mixture for secondary processing was observed.
 - No difference in purity, XRPD, DSC melting, ... of drug substance
 - Difference in appearance by scanning electron microscopy



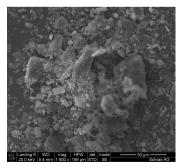
EXAMPLE (1-2): SEM – QUALITATIVE DESCRIPTION

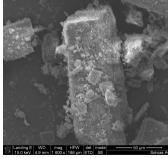






Various batches smaller scale
Used for DP process development



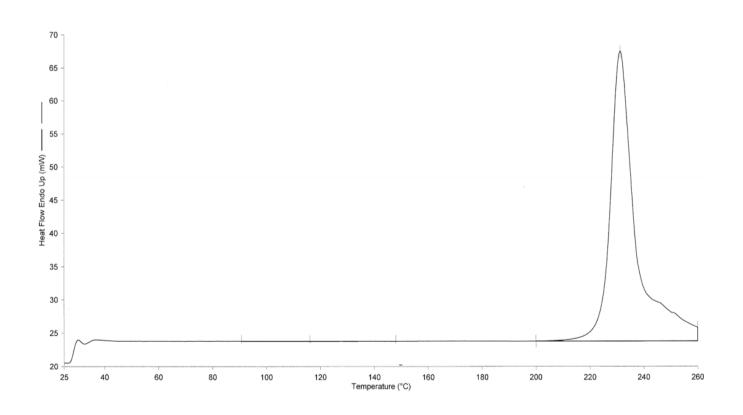




Various batches large scale Not processable – poor flow

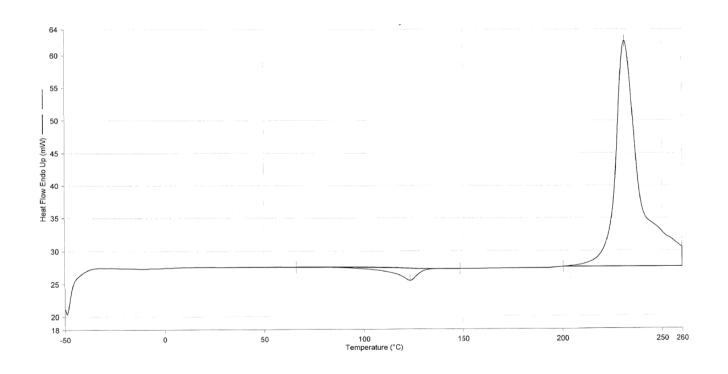


EXAMPLE (1-3): DSC OF REAL SAMPLE



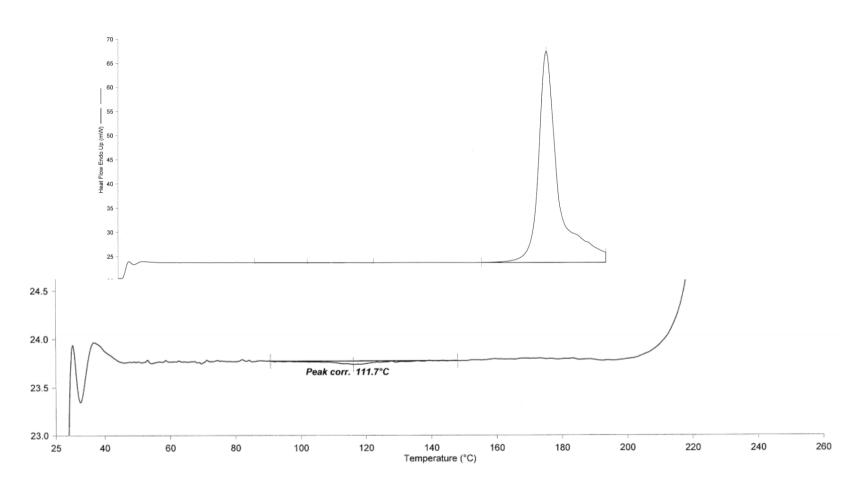


EXAMPLE (1-4): DSC OF SAMPLE GROUND IN MORTAR





EXAMPLE (1-5): DSC OF REAL SAMPLE - REVISITED



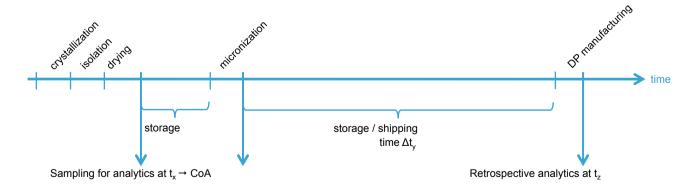


EXAMPLE (1-6): DSC – AGING OF A SAMPLE

► DSC results – tracing of exothermic event

t0m -1.5 J/g \rightarrow poor flow t3m -0.5 J/g \rightarrow flow ok t8m < LOD

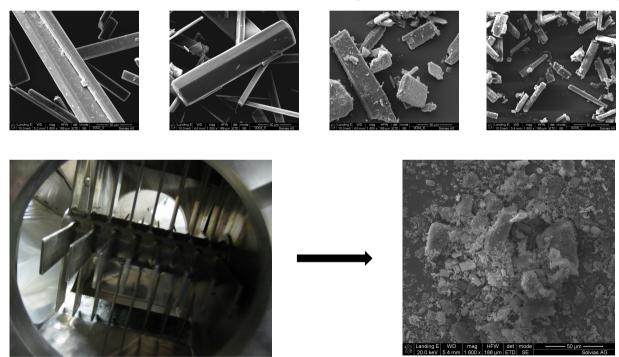
Remember DS as function of time!





EXAMPLE (1-7): ROOT CAUSE ANALYSIS:

Check critical unit operations – crystallization, isolation, drying, etc.

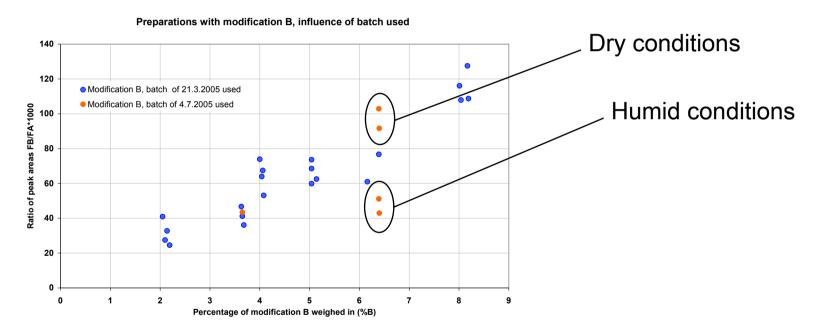


Delumping before filling drums was critical unit operation!



METHOD DEVELOPMENT: POSSIBLE PITFALLS

Polymorphic purity in DS: Case study by XRPD



- → Form B to form A conversion under moist conditions is accelerated
- → Consider kinetic stability upon all steps (mixture preparation, etc.)

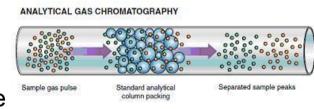


SOPHISTICATED TECHNIQUES: IGC

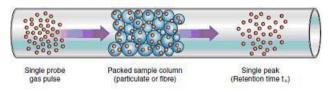
- ▶ iGC: Inverse Gas Chromatography
 - Drug substance is packed to a column
 - Single probe gas is used for probing the surface
 - Probe gas is changed
- ► Two operation modes
 - At infinite dilution
 - At finite concentration

Literature

Ho et al., 'Role of Surface Chemistry and Energetics in High Shear Wet Granulation' Ind. Eng. Chem. Res. 50, 2011, 9642-9649



INVERSE GAS CHROMATOGRAPHY (IGC)





SOPHISTICATED TECHNIQUES: IGC-ID (INFINITE DILUTION)

- ▶ During the travel through the column, the injected probes are interacting with the sample surface leading to numerous cycles of molecules adsorption-desorption.
- Several probes, having different properties, are injected separately at very low concentration (no interaction between the injected probes).
- The multiplicity of the collected ΔG_a allows the determination of the surface energy (γ_s^d) , the surface nanoroughness and of the surface acid-base character.



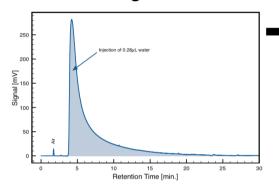
SOPHISTICATED TECHNIQUES: IGC-FC (FINITE CONCENTRATION)

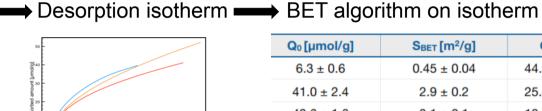
- ▶ IGC at Finite Concentration (IGC-FC) measurements involve important amounts of probe, leading to a large surface coverage, close or above to the monolayer coverage.
- ▶ IGC-FC offers the convenient way to record the desorption isotherm of a multitude of solutes in a large range of measurement temperatures.
- ► The interpretation of the isotherms leads to information such as specific surface area and the evaluation of the surface energetic heterogeneity (Adsorption energy distribution function).



SOPHISTICATED TECHNIQUES: IGC-FC (FINITE CONCENTRATION)

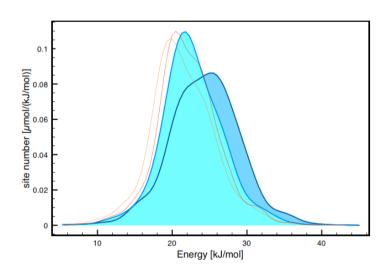
Chromatogram of a batch, e.g. with water







AEDF algorithm on isotherm



- Differences made visible for different DS batches
- Significance?
- No correlation with wet granulation performance observed



CHALLENGES (I)

- ▶ What is to be tracked?
- ▶ What technique?
- ▶ What method?
- What sophistication?
- ► Significance?



CHALLENGES (II) – PARTIAL ANSWER TO QUESTION

- Understand your drug substance solid state behavior, as function of:
 - Temperature
 - Relative humidity
- Know the unit operations in drug substance manufacturing, what changes if site/ scale is changed
- Understand the secondary processing, drug product manufacturing
- ⇒ Each of this point is a challenge at early stages of projects and might change over time
- → The more data is available and the more the rational of DP development is known the better potential problems can be anticipated



CHALLENGES (III) – GENERAL

- Documentation and description is key
 - It does not work' or 'it is different' is not sufficient
- ▶ If various labs (and organizations) are involved, be sure about common understandings and methods (e.g., particularly true for laser diffraction)
- Particular attention to sampling and sample preparation!
- External conditions (what is ambient? r.h. / T)
- Time development time axis as well as aging



THANK YOU.

